

Fundamental Physical Constants — Frequently used constants

Quantity	Symbol	Value	Unit	Relative std. uncert. u_r
speed of light in vacuum	c, c_0	299 792 458	m s^{-1}	exact
magnetic constant	μ_0	$4\pi \times 10^{-7}$	N A^{-2}	
		$= 12.566 370 614\dots \times 10^{-7}$	N A^{-2}	exact
electric constant $1/\mu_0 c^2$	ϵ_0	$8.854 187 817\dots \times 10^{-12}$	F m^{-1}	exact
Newtonian constant of gravitation	G	$6.674 08(31) \times 10^{-11}$	$\text{m}^3 \text{kg}^{-1} \text{s}^{-2}$	4.7×10^{-5}
Planck constant	h	$6.626 070 040(81) \times 10^{-34}$	J s	1.2×10^{-8}
$h/2\pi$	\hbar	$1.054 571 800(13) \times 10^{-34}$	J s	1.2×10^{-8}
elementary charge	e	$1.602 176 6208(98) \times 10^{-19}$	C	6.1×10^{-9}
magnetic flux quantum $h/2e$	Φ_0	$2.067 833 831(13) \times 10^{-15}$	Wb	6.1×10^{-9}
conductance quantum $2e^2/h$	G_0	$7.748 091 7310(18) \times 10^{-5}$	S	2.3×10^{-10}
electron mass	m_e	$9.109 383 56(11) \times 10^{-31}$	kg	1.2×10^{-8}
proton mass	m_p	$1.672 621 898(21) \times 10^{-27}$	kg	1.2×10^{-8}
proton-electron mass ratio	m_p/m_e	1836.152 673 89(17)		9.5×10^{-11}
fine-structure constant $e^2/4\pi\epsilon_0\hbar c$	α	$7.297 352 5664(17) \times 10^{-3}$		2.3×10^{-10}
inverse fine-structure constant	α^{-1}	137.035 999 139(31)		2.3×10^{-10}
Rydberg constant $\alpha^2 m_e c / 2h$	R_∞	10 973 731.568 508(65)	m^{-1}	5.9×10^{-12}
Avogadro constant	N_A, L	$6.022 140 857(74) \times 10^{23}$	mol^{-1}	1.2×10^{-8}
Faraday constant $N_A e$	F	96 485.332 89(59)	C mol^{-1}	6.2×10^{-9}
molar gas constant	R	8.314 4598(48)	$\text{J mol}^{-1} \text{K}^{-1}$	5.7×10^{-7}
Boltzmann constant R/N_A	k	$1.380 648 52(79) \times 10^{-23}$	J K^{-1}	5.7×10^{-7}
Stefan-Boltzmann constant $(\pi^2/60)k^4/\hbar^3c^2$	σ	$5.670 367(13) \times 10^{-8}$	$\text{W m}^{-2} \text{K}^{-4}$	2.3×10^{-6}
Non-SI units accepted for use with the SI				
electron volt (e/C) J	eV	$1.602 176 6208(98) \times 10^{-19}$	J	6.1×10^{-9}
(unified) atomic mass unit $\frac{1}{12}m(^{12}\text{C})$	u	$1.660 539 040(20) \times 10^{-27}$	kg	1.2×10^{-8}